## <u>Kurukshetra University Kurukshetra</u> <u>Scheme and Syllabi</u>

## for B.Sc. Part-I, Part-II and Part-III (Chemistry) w.e.f. 2014-2015, 2015-2016 and 2016-2017

#### **B.Sc. Part-I (Ist Semester)**

Paper	Code No.	Nomenclature	Max. Marks	Time
No.			Written + $I.A^{\#}$ .	
I	CH-101	Inorganic Chemistry (Th)*	32 + 8	3 hrs.
II	CH-102	Inorganic Chemistry (Th)	32 + 8	3 hrs.
III	CH-103	Organic Chemistry (Th)	32 + 8	3 hrs.

<sup>\*</sup>Th means Theory, # I.A. means Internal Assessment.

### **B.Sc. Part-I (IInd Semester)**

Paper	Code No.	Nomenclature	Max. Marks	Time
No.			Written + I.A.	
IV	CH-104	Inorganic Chemistry (Th)	32 + 8	3 hrs.
V	CH-105	Physical Chemistry (Th)	32 + 8	3 hrs.
VI	CH-106	Organic Chemistry (Th)	32 + 8	3 hrs.
VII	CH-107	Practicals	60	7 hrs.
			( No Internal	
			Assessment	
			in Practical	
			Exam)	

Note: Practical Exam will be held at the end of  $2^{nd}$  Semester Total Marks of I & II Semesters = 120 + 120 + 60 = 300

### **B.Sc. Part-II (IIIrd Semester)**

Paper	Code No.	Nomenclature	Max. Marks	Time
No.			Written + I.A.	
VIII	CH-201	Inorganic Chemistry (Th)	32 + 8	3 hrs.
IX	CH-202	Physical Chemistry (Th)	32 + 8	3 hrs.
X	CH-203	Organic Chemistry (Th)	32 + 8	3 hrs.

**B.Sc. Part-II (IVth Semester)** 

Paper	Code No.	Nomenclature	Max. Marks	Time
No.			Written + I.A.	
XI	CH-204	Inorganic Chemistry (Th)	32 + 8	3 hrs.
XII	CH-205	Physical Chemistry (Th)	32 + 8	3 hrs.
XIII	CH-206	Organic Chemistry (Th)	32 + 8	3 hrs.
XIV	CH-207	Practicals	60	7 hrs.
			( No Internal	
			Assessment in	
			Practical Exam)	

Note: Practical Exam will be held at the end of 4th Semester Total Marks of III & IV Semesters = 120 + 120 + 60 = 300

**B.Sc. Part-III (Vth) Semester** 

Paper	Code No.	Nomenclature	Max. Marks	Time
No.			Written + I.A.	
XV	CH-301	Inorganic Chemistry (Th)	32 + 8	3 hrs.
XVI	CH-302	Physical Chemistry (Th)	32 + 8	3 hrs.
XVII	CH-303	Organic Chemistry (Th)	32 + 8	3 hrs.

## **B.Sc. Part-III (VIth Semester)**

Paper	Code No.	Nomenclature	Max. Marks	Time
No.			Written + I.A.	
XVIII	CH-304	Inorganic Chemistry (Th)	32 + 8	3 hrs.
XIX	CH-305	Physical Chemistry (Th)	32 + 8	3 hrs.
XX	CH-306	Organic Chemistry (Th)	32 + 8	3 hrs.
XXI	CH-307	Practicals	60	7 hrs.
			( No Internal Assessment in	
			<b>Practical Exam</b> )	

Note: Practical Exam will be held at the end of 6<sup>th</sup> Semester

Total Marks of V & VI Semesters = 120 + 120 + 60 = 300TOTAL MARKS OF CHEMISTRY in B.Sc. degree =  $300 \times 3 = 900$ 

### B.Sc. I Year Paper-VII (CH-107) Practicals

M.Marks:60 Time: 7 Hrs.

#### (One day in two sessions i.e. 9:00AM-12:30PM and 1:30PM-5:00PM) Section-A (Inorganic)

#### **Volumetric Analysis**

- **1.** Preparation of reference solutions.
- **2. Redox titrations**: Determination of  $Fe^{2+}$ ,  $C_2O_4^{2-}$  (using KMnO<sub>4</sub>,  $K_2Cr_2O_7$ )
- **3. Iodometic titrations:** Determination of Cu<sup>2+</sup> (using standard hypo solution).
- **4.** Complexometric titrations: Determination of  $Mg^{2+}$ ,  $Zn^{2+}$  by EDTA.

#### Paper Chromatography

Qualitative Analysis of any one of the following Inorganic cations and anions by paper chromatography  $(Pb^{2+}$ ,  $Cu^{2+}$ ,  $Ca^{2+}$ ,  $Ni^{2+}$ ,  $Cl^-$ ,  $Br^-$ ,  $l^-$  and  $PO_4^{3-}$  and  $NO_3^{--}$ ).

#### **Section-B** (Physical)

- 1. To determine the surface tension of at least two liquids using stalagmometer by drop no. and drop weight methods (Use of organic solvents excluded).
- **2.** To study the effect of surfactant on surface tension of water.
- **3.** To determine the viscosity of at least two liquids by using Ostwald's viscometer (Use of organic solvents excluded).
- **4.** To determine the specific refractivity of at least two liquids.

## Section-C (Organic)

- 1. Preparation and purification through crystallization or distillation and ascertaining their purity through melting point or boiling point
  - (i) Iodoform from ethanol (or acetone)
  - (ii) m-Dinitrobenzne from nitrobenzene (use 1:2 conc.  $HNO_3$ - $H_2SO_4$  mixture if fuming  $HNO_3$  is not available)
  - iii) p-Bromoacetanilide from acetanilide
  - iv) Dibenzalacetone from acetone and benzaldehyde
  - v) 2,4-DNP derivative of Benzophenone/Acetophenone.
- 2. To study the process of (i) sublimation (ii) Crystallization of camphor and phthalic acid

#### Distribution of marks

1.	Section-A	15 marks
2.	Section-B	15 marks
3.	Section-C	15 marks
4.	Viva-voce	05 marks
5.	Lab Record	10 marks

#### B.Sc. II Year Paper XIV (CH-207) Practicals

M.Marks:60 Time: 7 Hrs. (One day in two sessions i.e. 9:00AM-12:30PM and 1:30PM-5:00PM)

#### Section-A (Inorganic)

#### 1. Gravimetric Analysis:

Quantitative estimations of, Cu<sup>2+</sup> as copper thiocyanate, Ni<sup>2+</sup> as Ni – dimethylglyoxime and Al<sup>3+</sup>as oxinate.

#### 2. Colorimetry:

To verify Beer - Lambert law for  $KMnO_4/K_2Cr_2O_7$  and determine the concentration of the given  $KMnO_4/K_2Cr_2O_7$  solution.

3.**Preparations:** Preparation of Cuprous chloride, tetra ammine cupric sulphate, chrome alum, potassium trioxalatochromate(III) and Nickel Hexammine chloride.

#### **Section-B** (Physical)

- 1. To determine the CST of phenol water system.
- 2. To determine the solubility of benzoic acid at various temperatures and to determine the ▲ H of the dissolution process.
- 3. To determine the enthalpy of neutralisation of a weak acid/weak base vs. strong base/strong acid and determine the enthalpy of ionisation of the weak acid/weak base.
- 4. To determine the enthalpy of solution of solid calcium chloride.
- 5. To study the distribution of iodine between CCl<sub>4</sub> and water.
- 6. Determine rate constant of hydrolysis of CH<sub>3</sub>COOC<sub>2</sub>H<sub>5</sub>.

### Section-C (Organic)

Systematic identification (detection of extra elements, functional groups, determination of melting point or boiling point and preparation of at least one pure solid derivative) of the following simple mono and bifunctional organic compounds: Naphthalene, anthracene, acenaphthene, benzyl chloride, p-dichlorobenzene, m-dinitrobenzene, p-nitrotoluene, resorcinol , hydroquinone,  $\alpha$ -naphthol,  $\beta$ -naphthol, benzophenone, ethyl methyl ketone, benzaldehyde, vanillin, oxalic acid, succinic acid, benzoic acid, salicyclic acid, aspirin, phthalic acid, cinnamic acid, benzamide, urea, acetanilide, benzanilide, aniline hydrochloride, p-toluidine, phenyl salicylate (salol), glucose, fructose, sucrose, o-, m-, p-nitroanilines, thiourea.

#### Distribution of marks

1.	Section A	15 marks
2.	Section B	15 marks
3.	Section C	15 marks
4.	Viva-voce	05marks
5.	Lab Record	10 marks

#### B.Sc. III Year Paper-XXI (CH-307) Practicals

Max.Marks:60 Time: 7 Hrs.

#### (One day in two sessions i.e. 9:00AM-12:30PM and 1:30PM-5:00PM) Section-A (Inorganic)

Semimicro qualitative analysis of mixture containing not more than four radicals (excluding interfering, Combinations and insoluables):

#### Section-B (Physical)

- 1. To determine the strength of the given acid solution (mono acid only) conductometrically.
- 2. To determine the solubility and solubility product of a sparingly soluble electrolyte conductometrically.
- 3. To determine the strength of given Ferrous ammonium sulphate solution potentiometrically.
- 4. To determine the molecular weight of a non-volatile solute by Rast method.
- 5. Prepration of acidic and basic buffers and comparison of their pH with theoretical values.
- 6. To determine the specific rotation of optically active substance (any two).

#### **Section-C (Organic)**

#### 1. Thin Layer Chromatography

(Determination of  $R_f$  values and identification of organic Compounds)

Separation of a mixture of coloured organic compounds using common organic solvents.

- 2. To separate the binary liquid mixtures using distillation.
- 3. Synthesis of the following organic compounds:
  - (a) To prepare salicylic acid from Aspirin.
  - (b) To prepare p-bromoaniline from p-bromoacetanilide.
  - (c) To prepare m-nitroaniline from m-dinitrobenzene.
  - (d) To prepare S-Benzyl-iso-thiouronium chloride from Thiourea.

#### **Distribution of marks**

1.	Section- A	15 marks
2.	Section-B	15 marks
3.	Section-C	15 marks
4.	Viva-voce	05 marks
5.	Lab Record	10 marks

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## IGN COLLEGE, LADWA

## **DEPARTMENT OF CHEMISTRY**

## **ATTENDACE CHART**

**OF** 

**UNIVERSITY PRACTICAL EXAMINATIONS** 

# KURUKSHETRA UNIVERSITY, KURUKSHETRA RISC TIL YEAR EXAMINATION 200 April 2019

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Subject CHEMISTRY (CHI-307) Date 17.64.201

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(Established by the State Legislature Act XII of 1956)

B. Sc. 711 Year EXAMINATION, 200 April 2019

### PRACTICAL ATTENDANCE CHART

Chart for Preparing List of each grou	p for P	racticals in	Science and	other Subjects
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## KURUKSHETRA UNIVERSITY, KURUKSHETRA (Established by the State Legislatura Act XII of 1956) B. Sc. III. Year EXAMINATION, 200 April 2017 PRACTICAL ATTENDANCE CHART

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(Established by the State Legislature Act XII of 1956)

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to the Deputy/Assit Registrar (Secrecy) alongwith Award-List 3 (42)(5)

#### KEROKSHETRA UNIVERSITY, KURUKSHETRA (Established by the State Legislature Act XII of 1956) Se II EXAMINATION, 200 Amil 2019 PRACTICAL ATTENDANCE CHART ach group for Practicals in Science and other Subjects in I. G. N. College W. Du A - dx 25-04-9.01 HM -\_ Date \_\_ Time 9 00 MM — 12 30 MM 1.30 PM — 5 00 PM gnature of the Candidate before Signature of the Candidate before the Practical Examiner Priyanka 170042901 Biyanka Kiparka 170042902 Polnec Prince Poince 170042903 Anisha Anustra rinusha 170062904 Anchal Anchal duralla 1700 12905 Anwadha omegadhe 17 ou 42906 Deepak James Leepale kuman 17 0042907 frisher Josmeet Kaus Jasme et Kay 170042909 Jasmeet Kaus 17 0049 910 DeeRika Deepika DeeRika Sunaypurt la 170042911 Somarpaired Kour Samarkurelkaus 170042913 Sont Son Sond Trope Jyoti Sachin Larni 121 Jyoti 170042915 Sachen Saini Paras Hardhan Sachin Saini Paras Mandian Paras Mandhan 170042916 14. Juati Devi Swell Devi 170042918 Swell Devi 170042919 ABSENT 16 falel fall 1700 6292 falil Blubban Paw. 1707 42922 Shubban Parul hubbram Punia Robin 1700 42924 Robin Robin Kanchah Kanchan 170042926 Kanchan 1700 42927 ABSENT kamal Ran. Komal Kani (coma) Rani 17 0042928 Total No. of Candidates examined by Total No of Candidates absent in the Practical Examination 02 Signature of Practical Examiner Address Prof. Vi Krom Kumar Address Prof. Vi Krom Kumar Deptt of Chem. TB College Panipal Signature & Address of the Co-Examinational Dr. R. K. Chauhay 20 the Examiner 20 Total No. of Candidates absent in the

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(Established by the State Legislature Act XII of 1956)

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